

*Annual Review of Biochemistry***PI(4,5)P₂ Clustering and Its
Impact on Biological Functions**

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Annu. Rev. Biochem. 2021. 90:681–707

First published as a Review in Advance on
January 13, 2021The *Annual Review of Biochemistry* is online at
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Keywordsphosphatidylinositol 4,5-bisphosphate, PI(4,5)P₂, cluster, multivalent cations, viral proteins, plasma membrane**Abstract**

Located at the inner leaflet of the plasma membrane (PM), phosphatidylinositol 4,5-bisphosphate [PI(4,5)P₂] composes only 1–2 mol% of total PM lipids. With its synthesis and turnover both spatially and temporally regulated, PI(4,5)P₂ recruits and interacts with hundreds of cellular proteins to support a broad spectrum of cellular functions. Several factors contribute to the versatile and dynamic distribution of PI(4,5)P₂ in membranes. Physiological multivalent cations such as Ca²⁺ and Mg²⁺ can bridge between PI(4,5)P₂ headgroups, forming nanoscopic PI(4,5)P₂-cation clusters. The distinct lipid environment surrounding PI(4,5)P₂ affects the degree of PI(4,5)P₂ clustering. In addition, diverse cellular proteins interacting with PI(4,5)P₂ can further regulate PI(4,5)P₂ lateral distribution and accessibility. This review summarizes the current understanding of PI(4,5)P₂ behavior in both cells and model membranes, with emphasis on both multivalent cation- and protein-induced PI(4,5)P₂ clustering. Understanding the nature of spatially separated pools of PI(4,5)P₂ is fundamental to cell biology.

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1. PHOSPHOINOSITIDES

Phosphoinositides (PIPs) are phosphorylated derivatives of phosphatidylinositol (PI) in cells. Since the discovery of PI/PIP signaling cycles by Mabel and Lowell Hokin in the 1950s (1), these lipids have gained significant research interest because they profoundly affect hundreds of biochemical processes in eukaryotic cells. PI is synthesized primarily in the endoplasmic reticulum (ER) and is then transferred to other membranes either by vesicular transport or by nonvesicular lipid transport via PI transfer proteins (PITPs) (2). PITPs sequester PI from a membrane in a hydrophobic pocket and deliver it to specific membrane compartments for further phosphorylation by numerous lipid kinases, which makes PITPs critical regulators of PIP pathways (3, 4). Reversible phosphorylation of the PI inositol ring at positions 3, 4, and 5 produces seven PIP species. Each PIP is predominantly found at a distinct subcellular location (**Figure 1**). Their spatial distribution, steady-state levels, and conversion to other species are primarily regulated by PIP kinases and phosphatases, whose localization and activation are also tightly controlled (5). Among PIPs, phosphatidylinositol 4,5-bisphosphate [PI(4,5)P₂] is the most abundant PIP found in mammalian cells. PI(4,5)P₂ plays critical roles in cell life and death, and many aspects of PI(4,5)P₂ biology have been intensively reviewed (6–13). The *in vivo* effects of PI(4,5)P₂ are complicated to work out because hundreds of proteins bind to it. In addition, PI(4,5)P₂ itself is an unusual phospholipid with unique properties that complicate its biochemistry, and these complexities likely affect its behavior in cells, even independently of protein binding. This review focuses on the biochemical and biophysical properties of PI(4,5)P₂ and highlights its unique clustering properties and its effects on biological functions.

2. PI(4,5)P₂ HEADGROUPS AND ACYL CHAINS

PI(4,5)P₂ is among the most highly charged anionic phospholipids, and it exhibits rich physical chemistry. PI(4,5)P₂ contains two phosphomonoesters at positions 4 and 5 and one

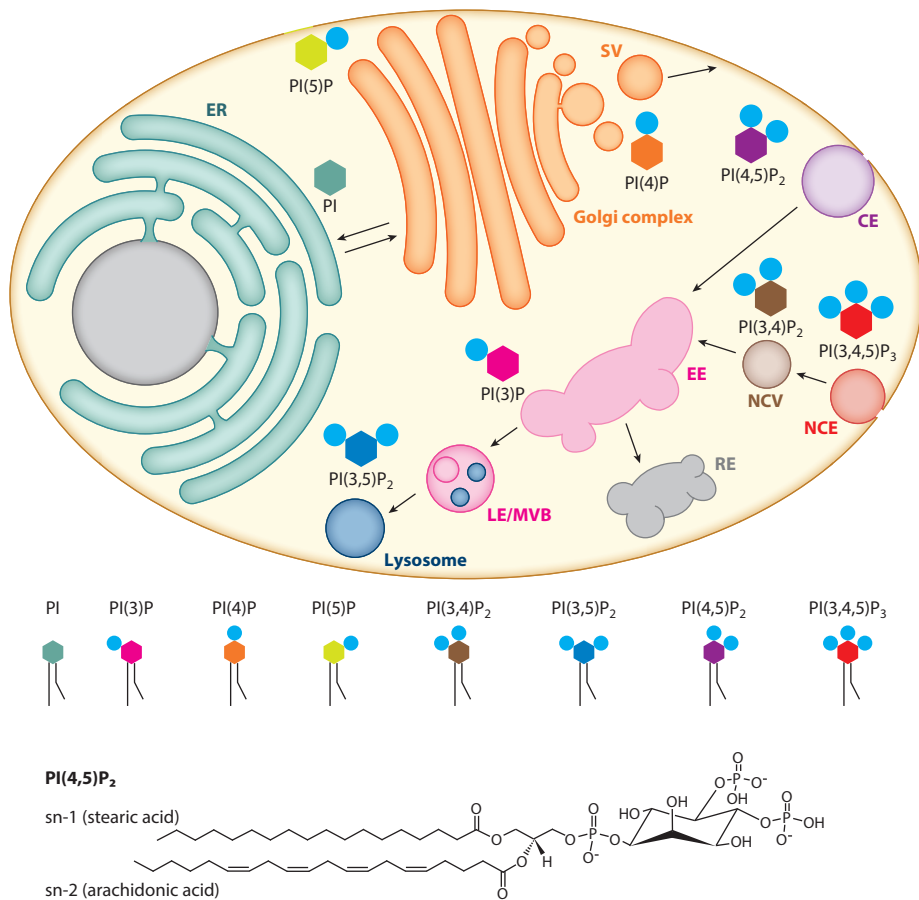


Figure 1

Subcellular distributions of PI and PIPs. The headgroup of each PIP is simplified as an inositol ring with 1–3 phosphate groups at different positions. This depiction is a simplification, illustrating only the most prominent location where each phosphoinositide species is found. Black arrows indicate the progression of the membrane trafficking pathway. The PI(4,5)P₂ structure is shown at the bottom. PI(4,5)P₂ containing stearic acid (18:0) on sn-1 and arachidonic acid (20:4) on sn-2 is the predominant species of natural PI(4,5)P₂. Abbreviations: CE, clathrin-dependent endocytosis; EE, early endosome; ER, endoplasmic reticulum; LE, late endosome; MVB, multivesicular body; NCE, nonclathrin endocytosis; NCV, nonclathrin endocytic vesicle; PI, phosphatidylinositol; PIP, phosphoinositides; RE, recycling endosome; SV, secretory vesicle.

phosphodiester at position 1 that connects the headgroup to the glycerol backbone (**Figure 1**). The ionization behavior of PI(4,5)P₂ has been studied extensively using ³¹P-nuclear magnetic resonance (NMR) (14–19). In micelles and bilayers, the charge of PI(4,5)P₂ was estimated based on the phosphorus chemical shift at different pH values (20, 21). To determine the contribution to the membrane charge from anionic lipids such as PIPs, it is important to know their intrinsic pK_a (22). The pK_as of the 4-phosphate and 5-phosphate were found to be 6.7 and 7.7, respectively, resulting in an overall charge of approximately –4 at a physiological pH of 7.2 (14). However, PI(4,5)P₂ in bilayers has a net charge of approximately –3 at the physiological ionic strength of 100 mM KCl at pH 7.0 according to electrophoretic mobility assays. This decreased charge per membrane-bound PI(4,5)P₂ is caused by the abundant phosphatidylserine (PS) in physiological

membranes, which renders the overall membrane surface charge negative, making the creation of additional negative charge from PI(4,5)P₂ headgroup ionization less favorable (21, 23). Electrostatic effects, including the electric field differences between PI(4,5)P₂ and PS, have been explored by McLaughlin et al. (10). PS and PI(4,5)P₂ can attract proteins with basic patches in a distinct manner and sometimes in a synergistic manner (24). Upon protein binding or during Ca²⁺ or Zn²⁺ transient influx, the proton bound to PI(4,5)P₂ might also be displaced; thus, the net charge of PI(4,5)P₂ could be -3, -4, or -5. Different ionization properties of the phosphomonoesters allow PI(4,5)P₂ to form both intramolecular hydrogen bonds with adjacent hydroxyl groups and intermolecular hydrogen bonds with neighboring PI(4,5)P₂ molecules (14, 15, 17–19).

Natural PI(4,5)P₂ has a mixture of different acyl chains, with the predominant species being sn-1 stearoyl and sn-2 arachidonoyl (25). The two acyl chains attached to the first and second carbons of the glycerol are denoted as sn-1 and sn-2, respectively. The saturated stearoyl chain is 18 carbons long, while the polyunsaturated arachidonoyl chain is 20 carbons long with four double bonds. This high degree of unsaturation makes PI(4,5)P₂ prefer a disordered environment (26). However, PI(4,5)P₂ has also been reported to be associated with membrane regions known as rafts or detergent-resistant membranes (10, 13, 27, 28), which are enriched in cholesterol and saturated phospholipids with a higher degree of membrane order than other membrane regions (29, 30). The association between PI(4,5)P₂ and the lipid rafts is a puzzling observation, because rafts are a property of the coexisting phases in the outer leaflet of the plasma membrane (PM), whereas PI(4,5)P₂ is exclusively located at the inner PM leaflet. A plausible explanation, but one not yet supported by experiments, is that the inner leaflet PI(4,5)P₂ is located preferentially across the PM from an outer leaflet raft, leading to the cytosolic PI(4,5)P₂-binding proteins being identified as raft markers (31).

3. PI(4,5)P₂ FUNCTION IN CELLS

PI(4,5)P₂ composes approximately 1–2 mol% of total PM lipids, corresponding to 10,000–20,000 molecules/μm² at the inner leaflet where it is found (32). In a crude model of all phospholipids being dissolved in the cytoplasm, the effective concentration of PI(4,5)P₂ in the cell was calculated to be approximately 10 μM (10). The synthesis of PI(4,5)P₂ is primarily mediated by the PIP kinases phosphatidylinositol 4-phosphate 5-kinase (PIP5K), which phosphorylates PI(4)P, or to a lesser extent, phosphatidylinositol 5-phosphate 4-kinase (PIP4K), which phosphorylates PI(5)P (33). PI(4,5)P₂ is critically involved in different signal transduction pathways that regulate a broad spectrum of biological activities. PI(4,5)P₂ itself can act directly as a messenger (34) or serve as a precursor in the generation of the important secondary messengers inositol trisphosphate (IP₃) and diacylglycerol (DAG), which are produced when PI(4,5)P₂ is hydrolyzed by phospholipase C (PLC) upon receptor stimulation (35). The rapid recovery of PI(4,5)P₂ levels at the PM following receptor-induced hydrolysis suggests a feedback loop of PI(4,5)P₂ consumption and replenishment to ensure cell homeostasis and signaling. Several proteins, such as Nir2, E-Syt1, and STIM1, which are localized to ER-PM junctions, have been reported to replenish PI(4,5)P₂ at the PM and Ca²⁺ in the ER following receptor-induced cell signaling (36). Phosphatidylinositol 3,4,5-trisphosphate [PI(3,4,5)P₃], which is involved in cell survival, cell growth, intracellular vesicle trafficking, cytoskeletal rearrangement, and cell metabolism, is generated from PI(4,5)P₂ by class I phosphoinositide 3-kinases (PI3Ks) (37). Even though the majority of PI(4,5)P₂ is located at the PM, a substantial pool of PI(4,5)P₂ is also found in the nucleus. Nuclear PI(4,5)P₂ is proposed to be located within nonmembrane structures, stabilizing nuclear PI(4,5)P₂-binding proteins, thereby contributing to nuclear PI(4,5)P₂ signaling (38). PI(4,5)P₂ also plays central roles in a broad spectrum of cellular functions, including exocytosis and endocytosis (8, 39), ion channel

and transporter regulation (40, 41), actin cytoskeleton assembly (42), endosomal trafficking (43, 44), membrane fusion (45), cell polarization (46), phagocytosis (47), and cell directional migration (48). PI(4,5)P₂ downstream signaling and PI(4,5)P₂-derived metabolites have been implicated in human diseases such as diabetes, bipolar disorder, and cancer (49, 50).

With its turnover being spatially and temporally regulated, PI(4,5)P₂ is well suited to recruit diverse cellular proteins and to interact with various effector proteins. Typically, PI(4,5)P₂-binding proteins are classified by function into several major categories, including membrane transport and trafficking, actin cytoskeletal dynamics and organization, functional enzymes, PM binding, small GTPase guanine nucleotide exchange factors and GTPase-activating proteins, and protein kinases and phosphatases, and several minor categories, including cell adhesion molecules, transcription and translation factors, and microtubule proteins (51).

A large group of PI(4,5)P₂-binding proteins is involved in membrane transport and trafficking. Endocytosis is regulated by PI(4,5)P₂ at several steps including vesicle formation, maturation, and fission. Clathrin-mediated endocytosis (CME) is a well-studied form of endocytosis that involves numerous proteins interacting with PI(4,5)P₂ using different strategies (52). The pleckstrin homology (PH) domain in dynamin-1 and -2, the ENTH (Epsin N-terminal homology) domain of Epsin1 and 2, the Hip1/1R domain, and the ANTH (AP180 N-terminal homology) domain of AP180 and CALM interact with PI(4,5)P₂ at defined binding sites. The α and μ 2 subunits of AP-2, profilin, and the Wiskott-Aldrich Syndrome protein (WASP) family (44) bind to PI(4,5)P₂ via surface-exposed basic patches. Other actin-associated proteins, such as Bin-Amphiphysin-Rvs (BAR)-domain proteins that deform membranes, inducing protrusions or invaginations, interact with PI(4,5)P₂ headgroups electrostatically and cooperatively (53). Some BAR proteins insert amphipathic α -helices into the lipid bilayer. Other proteins that are involved in exocytosis associate with PI(4,5)P₂, including synaptotagmin-1, syntaxin-1A, Munc13, Rabphilin, VAMP-2, and granuphilin (9).

PI(4,5)P₂ is the best-characterized actin cytoskeleton regulator among all PIPs. PI(4,5)P₂ has been shown to facilitate actin cytoskeleton formation beneath the PM, and any change in PI(4,5)P₂ local concentrations affects actin dynamics. An increase in PI(4,5)P₂ levels activates proteins that induce actin filament assembly, such as ezrin, radixin, and moesin (ERM) family proteins, talin, and WASP family proteins that activate the Arp2/3 complex (42, 54). An increase in PI(4,5)P₂ levels also inhibits proteins that promote actin filament disassembly, such as gelsolin, heterodimeric capping protein, ADF/cofilin, profilin, and twinfilin (54). Actin and cytoskeletal modulating proteins use either defined PI(4,5)P₂-binding domains or a cluster of positively charged and/or hydrophobic residues such as MARCKS (myristoylated alanine-rich C kinase substrate), GAP43 (growth-associated protein of 43 kDa), and CAP23 (cortical-associated protein of 23 kDa, a cytoskeleton-associated protein). Furthermore, a number of actin-associated proteins, such as BAR-domain proteins, are capable of directly deforming PIP-rich membranes to induce PM protrusions or invaginations (53). A cluster of proteins targeted to the PM by PI(4,5)P₂ are classified as small GTPases or GTPase regulators, such as members of the Rho, Arf, Ras, and Rab families. These proteins usually harbor a polybasic patch and hydrophobic modifications such as palmitoyl, prenyl, or myristoyl. Proteins in the Rho and Arf6 families bind to PI(4,5)P₂ during cytoskeleton regulation (51).

PI(4,5)P₂ interacts directly with a variety of structured protein modules such as PH domains (55). PH domains, approximately 120 amino acids long, are best known for their ability to bind PIPs with high affinity and specificity, although less than 10% of all PH domains share this property (56). Most PH domains bind to PI(4,5)P₂ weakly and nonspecifically. Many PH domains share similar core structures consisting of a pair of antiparallel β -sheets with a C-terminal α -helix. There are three primary types of PIP-binding sites in PH domains: those that contain the KX_n(K/R)XR

motif or a canonical PIP-binding site in the loop connecting strands $\beta 1$ and $\beta 2$, such as a phospholipase C $\delta 1$ PH domain (PH-PLC $\delta 1$) (57, 58); those that do not harbor the $KX_n(K/R)XR$ motif and instead have a noncanonical PIP-binding site, as found in the β -spectrin PH domain (59); and those that contain both canonical and noncanonical PIP-binding sites, such as the ASAP1 PH domain (60, 61). PH-PLC $\delta 1$ is the best characterized of the PI(4,5) P_2 -specific binding proteins, and its fluorescent chimeric versions have been used to study cellular PI(4,5) P_2 localization and function (62, 63). Several basic residues in PH-PLC $\delta 1$ were predicted to bind to PI(4,5) P_2 (64). To test whether these basic residues are sufficient for specific PI(4,5) P_2 binding, a short peptide with amino acids corresponding to residues 30–43 of PH-PLC $\delta 1$ was synthesized; however, this peptide bound to PI(4,5) P_2 only weakly and exhibited little specificity (64). These observations suggest that the intact tertiary structure of the PH domain is required for high-affinity PI(4,5) P_2 -specific interaction. The formation of hydrogen bonding networks between 4- and 5-position phosphates and lysine (Lys)30 and Lys57 in PH-PLC $\delta 1$ locks PI(4,5) P_2 inside of the binding pocket (57). PH-PLC $\delta 1$ is reported to bind to the PI(4,5) P_2 polar headgroup with a 1:1 stoichiometry and a K_d of 2 μM in vitro (65). Additionally, PH-PLC $\delta 1$ also binds to the soluble D-myo-IP3 with even higher affinity, almost eightfold greater than its affinity for PI(4,5) P_2 (57). Excess IP3 can abolish the binding of the PH domain to PI(4,5) P_2 in vitro, which would interfere with the function of the PH domain as a PI(4,5) P_2 sensor in cells.

FERM domains, which are approximately 300 amino acids in length, were originally defined by their presence at the N terminus of the erythrocyte band 4.1 protein (F) and the related cytoskeletal proteins ezrin (E), radixin (R), and moesin (M) (66). These domains are also found in the cytoskeletal protein talin, the tumor suppressor merlin, and several tyrosine kinases (such as JAK and FAK) and phosphatases (such as PTPN3 and PTPN4) (67). Upon binding to PI(4,5) P_2 , the FERM domain of these proteins is released from its autoinhibitory state, allowing these proteins to serve as adaptors between the actin cytoskeleton and the PM (68). The FERM domain is composed of three subdomains, A, B, and C. Subdomain C contains a fold resembling a PH domain (69). However, the region responsible for PI(4,5) P_2 binding is a basic cleft between subdomains A and C. Further mutagenesis studies imply that other regions of the FERM domain may also contribute to PI(4,5) P_2 binding. The residues involved in PI(4,5) P_2 binding are conserved, and the sequence homology suggests they form a potential PI(4,5) P_2 -binding pocket.

Another PI(4,5) P_2 -binding module is the ENTH domain, which plays a critical role in CME (70). ENTH domains contain approximately 140 amino acids and are found in Epsin1 and 2 and Hip1R. NMR studies determined that positively charged residues in a cleft of the epsin ENTH domain contribute to PI(4,5) P_2 binding (71). ENTH binds to PI(4,5) P_2 much more strongly than a related domain called ANTH does. Another structured domain selectively binding to PI(4,5) P_2 is the PX domain of the CPK PI3K (70).

While many proteins have developed specific PI(4,5) P_2 -binding pockets, other proteins use unstructured basic regions to bind to PI(4,5) P_2 via nonspecific electrostatic interactions. The MARCKS protein is among the best studied of these (10, 72). MARCKS protein binds to the negatively charged inner leaflet of the PM using two mechanisms: An N-terminal myristate inserts hydrophobically into the bilayer, and a conserved effector domain interacts with the membrane via its 13 basic residues (residues 151–175: KKKKKRFSFKKSFKLSGFSFKKNNKK). The MARCKS effector domain binds to and sequesters three PI(4,5) P_2 molecules (10, 73). The five aromatic phenylalanine residues in the effector domain were shown to be inserted into bilayers (74–76). This insertion pulls the adjacent basic residues closer to the membrane surface where the electrostatic potential is stronger, thereby enhancing lateral PI(4,5) P_2 sequestration. Membrane-bound MARCKS can be dissociated from the PM by Ca^{2+} -loaded calmodulin or by phosphorylation of the Ser residues of the effector domain by protein kinase C (PKC) (10).

Treating these membrane-bound molecules simplistically, as if they were water-soluble molecules within the volume of the cytosol, the physiological level of MARCKS is estimated to be $\sim 2 \mu\text{M}$ in fibroblasts and $\sim 10 \mu\text{M}$ in neuronal tissue, a range similar to that estimated for PI(4,5)P₂ in cells (10). Thus, MARCKS could potentially bind and sequester a significant fraction of cellular PI(4,5)P₂. MARCKS-sequestered PI(4,5)P₂ would be less susceptible to hydrolysis by PLC δ 1 compared with exposed PI(4,5)P₂ when MARCKS is dissociated. In addition to MARCKS, PI(4,5)P₂ binds strongly to GAP43 and CAP23 (26). These three motility-associated proteins are termed GMC proteins, and they share important properties: They all contain a basic effector domain, are reported to localize in the cholesterol-enriched domains termed rafts, and laterally sequester PI(4,5)P₂ (77). The association between GAP43 and rafts was found to require dual palmitoylation of its N terminus (78).

4. ROLE OF PI(4,5)P₂ IN VIRUS REPLICATION

Besides the numerous roles PI(4,5)P₂ plays in maintaining normal cell functions, it is also indispensable during the life cycle of many viruses, including human immunodeficiency virus 1 (HIV-1) and Ebola virus. Consistent with its roles in vesicle transport, trafficking, and actin cytoskeletal regulation, PI(4,5)P₂ also affects viruses that hijack cellular machinery during various stages of infection.

The role of PI(4,5)P₂ begins as early as viral entry. Many enveloped viruses such as influenza virus and vesicular stomatitis virus use PI(4,5)P₂-dependent CME for productive infection (79, 80). Some nonenveloped viruses that use integrins as their receptors, such as foot-and-mouth disease virus, also are highly dependent on PI(4,5)P₂ for internalization (81). One study (82) reported that PI(4,5)P₂ plays a role in HIV-1 entry.

PI(4,5)P₂ is essential for the genome replication of some viruses, such as hepatitis C virus (HCV). HCV, like other flaviviruses, replicates its RNA genome in close contact with invaginated membranous structures that might be derived from organelles such as the ER, Golgi complex, and endosomes (83). The N-terminal amphipathic helix (AH) of the nonstructural protein NS5A binds to PI(4,5)P₂ through a pair of highly conserved basic amino acids, Lys20 and Lys26. This NS5A–PI(4,5)P₂ interaction induces a conformational change that stabilizes the interaction between NS5A and TBC1D20, a GTPase-activating protein for Rab1, which is required to establish efficient HCV replication (84). Importantly, these Lys residues are highly conserved across all HCV isolates and the NS4B protein of the distantly related polioviruses and rhinoviruses.

The most studied role of PI(4,5)P₂ during the viral life cycle is in viral assembly (85). Many cell studies have shown that PI(4,5)P₂ is critical for HIV-1 particle assembly (86, 87). The HIV-1 structural protein, Gag, is the primary driving force of viral assembly on membranes. Depleting PI(4,5)P₂ at the PM by overexpressing 5-phosphatase IV (5ptaseIV), which breaks down PI(4,5)P₂, significantly reduces both HIV-1 assembly at the PM and viral release from cells. Altering the localization of PI(4,5)P₂ to endosomal compartments by expressing an Arf6 mutant, Q67L, also drastically reduces virus release from cells (86). A recent study with living cells observed less HIV-1 Gag assembly following rapid PI(4,5)P₂ depletion at the PM and enhanced Gag assembly upon restoration of PM PI(4,5)P₂ levels (88). Similarly, several biochemical studies suggest that the membrane binding of HIV-1 Gag and Gag-related proteins is significantly enhanced in the presence of PI(4,5)P₂, as shown by both liposome flotation assays and pelleting assays (89, 90). All of these results suggest that basic residues in Matrix (MA), the membrane-binding domain of Gag, interact directly with PI(4,5)P₂. HIV-1 MA binds to membranes using a bipartite signal: The highly basic region binds to negatively charged lipids such as PS and PI(4,5)P₂ in membranes through electrostatic interaction, and the N-terminal myristate moiety inserts into the membrane core

through hydrophobic interactions. Specific HIV-1 MA interactions with PI(4,5)P₂ were detected by mass spectrometric protein footprinting (91). NMR studies also confirmed that the HIV-1 MA domain binds specifically to water-soluble short-chain (C4 and C8) analogs of PI(4,5)P₂ (91). Lipidomic studies suggest that HIV-1 virions have an elevated PI(4,5)P₂ level compared with the host PM from which the virions bud (92, 93). This PI(4,5)P₂ enrichment requires the HIV-1 MA domain; a Gag mutant lacking the polybasic globular head of MA but still containing the N-terminal myristate incorporates significantly less PI(4,5)P₂ than the wild type (94). HIV-1 viral membranes were also reported to have increased levels of the raft lipid components cholesterol and sphingomyelin, as well as lipids from the cytosolic leaflet, such as PI(4,5)P₂, PS, and plasmalogen-phosphatidylethanolamine (92–95). It was therefore hypothesized that HIV-1 buds from PI(4,5)P₂-rich raft-like microdomains and that HIV-1 exploits MA–PI(4,5)P₂ interactions for efficient virus assembly and release.

MA interaction with PI(4,5)P₂ has also been reported for other retroviruses, such as HIV-2, Mason-Pfizer monkey virus, equine infectious anemia virus, and murine leukemia virus (96–98). In contrast, human T-lymphotropic virus type 1 (99) and Rous sarcoma virus (RSV) (100) are reported to be much less dependent on PI(4,5)P₂ for viral assembly and release than HIV-1 Gag. In addition to retroviruses, several studies show that PI(4,5)P₂ is required to stabilize and/or induce oligomerization of the Ebola virus (a filovirus) structural protein VP40 at the PM. VP40 has been shown to be indispensable for viral particle formation (101). However, the VP40 of another filovirus, Marburg virus, seems to lack specific interactions with PI(4,5)P₂, instead acting as a promiscuous anionic charge sensor (102). A recent study suggests that influenza hemagglutinin (HA) protein tightly colocalizes with PI(4,5)P₂ at the PM in infected cells (103).

In addition to transmission through the release of viral particles into the extracellular space, viruses like HIV-1 can also be spread through cell–cell transmission by forming a virological synapse (VS) (104). A VS forms at the contact site of infected and uninfected T cells, mediated by envelope glycoprotein interactions with their receptor (CD4) and coreceptors (CXCR4 and CCR5) (105). Synapse complex formation relies on efficient actin cytoskeleton remodeling in the target cells (106). Upon Env and Gag recruitment to the cell–cell contact sites, viral particles transfer across the VS into the uninfected cells. PI(4,5)P₂ is involved in this process, since it regulates actin-binding protein accumulation that triggers VS formation, thus ensuring efficient viral dissemination.

5. PI(4,5)P₂ MICELLE FORMATION

The amphipathic nature of PI(4,5)P₂ results in two major forms of aggregation in the presence of water: micelles and membrane bilayers. These self-assemblies of phospholipids are driven by increased water entropy when the nonpolar tails are removed from water and are stabilized by electrostatic and hydrogen bond interactions among hydrophilic heads, cations, and the aqueous environment. Other phospholipids with long acyl chains form bilayers, but micelles are not detected because of extreme insolubility. The strongly hydrophilic headgroup makes PI(4,5)P₂ capable of forming micelles in an aqueous environment (107, 108), but in the presence of sufficient bilayer phase, PI(4,5)P₂ is more stable partitioning into the lipid bilayers.

A characteristic of any micelle-forming surfactant is the critical micelle concentration (CMC) (109), defined as the concentration above which surfactant monomers aggregate to form micelles. Beyond the CMC, any additional surfactant of the same type forms more micelles, leaving the monomer concentration almost constant. The shape and size of a micelle are dependent on the molecular structure and the aqueous conditions such as surfactant concentration, pH, ionic type and strength, and temperature. Among common methods to determine CMC are light scattering,

dye solubilization, and surface tension (109). Both light scattering and the solubility of a hydrophobic dye increase abruptly with increasing surfactant concentration above the CMC, whereas surface tension decreases with increasing surfactant monomer concentration, reaching a minimum value at the CMC. A few measurements of CMC values for PI(4,5)P₂ have been reported, for example, a CMC of 30–40 μM using light scattering (110), a CMC of 10 μM using a Coomassie blue dye method (111), and a CMC of 12.5 μM using the fluorescent dye DPH (112). Additionally, one group observed that the PI(4,5)P₂ CMC in the buffers 50 mM PIPES and 50 mM Tris at pH 7.0 is 30 μM, whereas that in water is 200 μM (113). Our data (Y. Wen, A. Matsumoto & G.W. Feigenson, unpublished data) suggest that the CMC of the natural brain PI(4,5)P₂ is approximately 0.5 μM in a buffer of 100 mM KCl, 20 mM HEPES, pH = 7.2, as measured by light scattering. We also found that the PI(4,5)P₂ CMC is lower in the presence of multivalent cations than in the presence of EDTA.

The aggregation number of each PI(4,5)P₂ micelle has been reported by gel filtration chromatography of the PI(4,5)P₂ micelle in complex with PI(4,5)P₂-binding proteins, such as PKC or profilin (114, 115). These complexes form micelles of molecular weight 93,000 with an estimated aggregation number of 82 and a Stokes radius of 39 Å. Interestingly, Janmey and colleagues (116, 117) showed that PI(4,5)P₂ exists as small micelles in the presence of buffer ions and/or monovalent cations, such as 100 mM NaCl or KCl, whereas the addition of millimolar concentrations of divalent cations such as Mg²⁺, Ca²⁺, or Ba²⁺ induces the formation of large, multilamellar PI(4,5)P₂ aggregates that are visible by electron microscopy (EM). This PI(4,5)P₂ cluster formation with divalent cations presumably occurs through bridging of PI(4,5)P₂ headgroups. The addition of EDTA or EGTA disrupts these PI(4,5)P₂ aggregates, confirming that this process is reversible. In summary, the formation of PI(4,5)P₂ aqueous micelles can be thought of as the form of PI(4,5)P₂ clustering behavior in the absence of bilayers. Even though this type of clustering as micelles is probably not physiologically relevant, it does indicate how interactive PI(4,5)P₂ headgroups are with each other, despite multiple negative charges.

6. PI(4,5)P₂ CLUSTERING BEHAVIOR IN VIVO

Evidence suggests that PI(4,5)P₂ forms clusters in the cellular PM. The Fujimoto group (118) visualized concentrated PI(4,5)P₂ pools at the rims of caveolae as well as in coated pits in cultured cells using an EM technique. Live cells were rapidly frozen without chemical fixation, and PI(4,5)P₂ was labeled with a GST-tagged PH-PLCδ1 probe. Caveolin-1 and caveolin-2 sequester PI(4,5)P₂ electrostatically, and these proteins assemble into heterooligomers to generate the caveolae with locally concentrated PI(4,5)P₂. Many studies report PI(4,5)P₂ clustering in intact cells or on membrane sheets derived from cells, and most of these studies detect PI(4,5)P₂ accumulation with fluorescently labeled PH-PLCδ1, acyl-chain-labeled fluorescent PI(4,5)P₂, or PI(4,5)P₂-specific antibodies. Many groups have used superresolution microscopy to visualize PI(4,5)P₂. Using direct stochastic optical reconstruction microscopy (dSTORM) with PI(4,5)P₂-specific antibodies directly conjugated with Alexa Fluor 647, PI(4,5)P₂ clusters were found in domains of approximately 65 nm in intact PC12 cells (119). Three quarters of the observed PI(4,5)P₂ clusters were reported to be elongated and one quarter to be circular. A study using stimulated-emission depletion (STED) microscopy suggested that PI(4,5)P₂ composes approximately 80% of the inner leaflet lipids within clusters. These PI(4,5)P₂ clusters measured approximately 70 nm in diameter when detected with fluorescent PH-PLCδ1 or approximately 90 nm in diameter when detected with a PI(4,5)P₂ antibody, which indicates that approximately 1,000 PI(4,5)P₂ molecules are present in each cluster (120). Another superresolution imaging study revealed that PI(4,5)P₂ distribution is homogenous in the majority of the PM by area, as detected by PH-PLCδ1. The authors

observed only some sparse PI(4,5)P₂-enriched regions with an average diameter of approximately 380 nm (121). Using fluorescence photoactivation localization microscopy (FPALM), a recent study showed that influenza HA protein coclusters with PI(4,5)P₂, with cluster diameters of approximately 135 nm and 182 nm with low and high influenza HA concentrations, respectively. Interestingly, the authors found that many PI(4,5)P₂ clusters are elongated instead of round and that PI(4,5)P₂-binding sites indicate a radial attractive gradient around the HA protein (103). Another experiment using FPALM with acyl-chain-labeled fluorescent PI(4,5)P₂ delivered across the PM by the carrier protein histone H1 also detected PI(4,5)P₂ clusters in live cells that were approximately 150 nm in diameter; even larger clusters are found in fixed cells, approximately 400 nm in diameter, when detected by fluorescent PH-PLC δ 1. Of note, another group using confocal microscopy (122) reported that the size of the PI(4,5)P₂ clusters detected using a PI(4,5)P₂ secondary antibody was enhanced upon adding a tertiary antibody. Thus, the visualization of PI(4,5)P₂ clusters using antibodies needs to be carefully examined and interpreted.

Numerous cellular PI(4,5)P₂-binding proteins are also proposed to modulate PI(4,5)P₂ distribution and induce PI(4,5)P₂ cluster formation. For example, MARCKS, GAP43, CAP23, and syntaxin-1 have been reported to laterally sequester PI(4,5)P₂ at the PM (77, 123). A large fraction of PI(4,5)P₂ at the inner leaflet is believed to be electrostatically sequestered or bound to and released by proteins containing a basic patch, such as MARCKS protein (10). Each MARCKS protein can bind three PI(4,5)P₂ molecules with a dissociation constant of 10⁻⁸ M (105). This bound PI(4,5)P₂ is released only upon transient Ca²⁺ influx or upon phosphorylation of a key serine residue by PKC. At the same time, other effector proteins could have access to these concentrated PI(4,5)P₂ platforms to activate downstream functions. Only a small fraction of PI(4,5)P₂ is believed to diffuse freely in the PM. The evidence for PI(4,5)P₂ being laterally sequestered by proteins comes from cellular studies. In several cell types, MARCKS is not uniformly distributed at the PM but is enriched in membrane ruffles or nascent phagosomes (124, 125). A similar PI(4,5)P₂ distribution pattern was later observed using fluorescent PH-PLC δ probes (125). Importantly, van Rheenen et al. (126), using advanced imaging methods to characterize the bumpy cell surface, showed that the apparent enrichment of PI(4,5)P₂ detected by PH-PLC δ clustering was not due to PI(4,5)P₂ cluster formation but instead was caused by an increased lipid content in submicroscopic folds and ruffles and thus was a visualization artifact. When MARCKS was disassociated from the PM into the cytosol by PKC phosphorylation, the free PI(4,5)P₂ level at the PM increased as membrane tension increased due to PI(4,5)P₂-dependent cytoskeletal adhesion to the PM. Another study used antibody labeling to show that PI(4,5)P₂ colocalized with the GMC proteins MARCKS, GAP43, and CAP23 in cholesterol-rich microdomains in many cell lines (77). Overexpression of GMC increases the detectable PI(4,5)P₂ clusters; GMC lacking effector domains seems to reduce PI(4,5)P₂ accumulation. However, the characteristic patchy pattern of GMC and PI(4,5)P₂ microdomains was not detectable when macroscopic clusters of GPI-linked proteins were prevented from forming by fixation with glutaraldehyde (77).

Other groups used unfixed PC12 cell membrane sheets, derived from the PM by gentle sonication, to study PI(4,5)P₂ lateral organization (120, 123). They reported that syntaxin-1, a SNARE protein that catalyzes regulated exocytosis, forms clusters in the PM, mediated by electrostatic interactions with PI(4,5)P₂. Syntaxin-1 contains a transmembrane domain juxtaposed to a polybasic region (basic residues KARRKK), which sequesters PI(4,5)P₂ headgroups. Using STED, they found that the polybasic region is required for PI(4,5)P₂ coclustering with syntaxin-1 and that nanoscopic PI(4,5)P₂ clusters were also required for syntaxin-1 sequestering (120, 123). PI(4,5)P₂-syntaxin clusters serve as recognition and association sites for vesicle docking during Ca²⁺-stimulated membrane fusion. Another group reported that Ca²⁺ links smaller PI(4,5)P₂-syntaxin-1 clusters into larger domains and that this clustering effect is reversible (127).

Locally concentrated PI(4,5)P₂-Ca²⁺ clusters can recruit more downstream effector proteins such as CAPS, Munc13, and synaptotagmin to facilitate assembly of the complete fusion machinery at the site of release (123). Together, these studies suggest that PI(4,5)P₂ clustering may occur as a consequence of lateral sequestration by effector proteins upon physiological Ca²⁺ influx and that these protein-PI(4,5)P₂ clusters regulate the accessibility of PI(4,5)P₂ to other effector proteins.

In addition to protein sequestration, upon activation of specific PIP kinases, PI(4,5)P₂ synthesis is localized to the inner leaflet of the PM (128). Type I PIP5K is mostly responsible for the generation of PI(4,5)P₂ from PI(4)P at the PM, since PI(4)P is the second most abundant PIP at the inner leaflet (7). Via distinct signaling pathways, PIP5K isoforms and splice variants have specific interactions with different proteins that allow PIP5K to target subcellular locations such as the PM, focal adhesions, the Golgi complex, and the nucleus (129). PI(4,5)P₂ is also locally synthesized at sites of actin remodeling, giving rise to membrane ruffles, filopodia, lamellipodia, and phagosomes. PI(4,5)P₂ accumulation at phagosomes due to PIP5K has been detected (43). This accumulation is transient, as the DAG level significantly increases and the PI(4,5)P₂ level decreases upon PLC activation. Small GTPases such as Rho, Rac family members, Arf6, specific proteins such as talin, and phospholipids such as phosphatidic acid (PA) have been shown to activate and regulate PIP5K activity and then modulate localized PI(4,5)P₂ synthesis (7, 9). Targeted PI(4,5)P₂ production by PIP5K further regulates various PI(4,5)P₂ effector proteins during biological functions. At a larger scale, compartmentalization in these small structures, such as in membrane protrusions of chemotactic and phagocytic cells, limits the rapid diffusion of locally generated PI(4,5)P₂ and can lead to a local PI(4,5)P₂ gradient (130). PI(4,5)P₂ gradients result from a combination of local synthesis, breakdown, and lipid lateral diffusion. Thus, it is likely that these separate PI(4,5)P₂ pools are spatially and temporally choreographed by a combination of enzymes, effector proteins, and probably other cytosolic factors (13). However, the mechanistic details of how the separate pools of PI(4,5)P₂ form are unclear, and *in vitro* studies could provide more information at a molecular level.

7. PI(4,5)P₂ CLUSTERING BEHAVIOR IN VITRO

Models of PI(4,5)P₂ accumulation *in vitro* are primarily based on mechanistic details of PI(4,5)P₂ cluster formation. Several models could explain PI(4,5)P₂ clustering in bilayers. PI(4,5)P₂ clustering is proposed to be induced by a combination of electrostatic bridging by multivalent cations such as Ca²⁺ and Mg²⁺ (**Figure 2**) (131–133), hydrogen bonding networks among PI(4,5)P₂ headgroups (10, 15, 17–19), and basic patches of proteins (74). Molecular dynamics simulations (134) indicate that clustered PI(4,5)P₂ affects protein binding, with protein binding further modulating PI(4,5)P₂ enrichment.

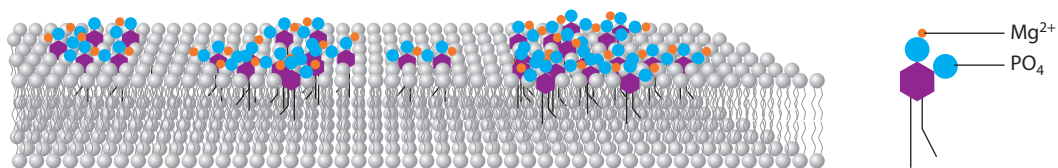


Figure 2

Schematic depiction of PI(4,5)P₂ clusters bridged by multivalent cations. Multivalent cations, such as Mg²⁺, are shown as orange circles. Each PI(4,5)P₂ headgroup is shown as a purple inositol ring with two blue phosphate groups at positions 4 and 5. PI(4,5)P₂ clusters form via electrostatic interactions between positively charged cations and negatively charged phosphate groups.

In the absence of multivalent cations or polybasic proteins, hydrogen bonding networks between PI(4,5)P₂ headgroups might induce PI(4,5)P₂ clustering despite electrostatic repulsion between highly negatively charged PI(4,5)P₂. Support for this hypothesis comes from the fact that demixing of PI(4,5)P₂ in fluid phosphatidylcholine (PC) bilayers can occur at membrane concentrations as low as 1% (14). Using Förster resonance energy transfer (FRET), PI(4,5)P₂ segregation from PC was observed at high pH, probably stabilized by hydrogen bond networks formed between the hydroxyl groups and the phosphomonoester and phosphodiester groups of adjacent PI(4,5)P₂. An NMR study (19) reported that PI(4,5)P₂ exhibited a biphasic pH-dependent ionization behavior, which could be explained by intermolecular sharing of the last remaining proton between the vicinal phosphomonoester groups. Strikingly, two studies (15, 18) showed that 5–20 mol% PI(4,5)P₂ induced macroscopic phase separation in giant unilamellar vesicles (GUVs) in the presence of 20 mol% PI or cholesterol while using multivalent cation-free buffers. PI was suggested to form a separate phase with PI(4,5)P₂, whereas cholesterol was observed to be present in both phases. The presence of such extremely large fractions of PI(4,5)P₂ in membranes and the absence of multivalent cations are not physiological conditions, so this kind of macroscopic phase separation might not occur at physiological PI(4,5)P₂ levels. It has also been reported that cholesterol, but not cholesterol derivatives with the hydroxyl group modified, promoted and stabilized PI(4,5)P₂ domains (15). This result indicates that the cholesterol hydroxyl group might participate in intermolecular hydrogen bond formation. Other bulk lipid components such as phosphatidylethanolamine (PE) could also influence PI(4,5)P₂ ionization properties, perhaps serving as hydrogen bond donors (17). A different group showed that the presence of chaotropic agents, such as monovalent salts, urea, and temperature, specifically and significantly expanded the PI(4,5)P₂ molecular area in lipid monolayers (135). These results indicate that without multivalent cations, hydrogen bond networks might cluster PI(4,5)P₂ and reduce the area per PI(4,5)P₂, forming condensed PI(4,5)P₂ clusters. Taken together, hydrogen bonding networks might contribute to PI(4,5)P₂ cluster formation; however, charge bridging by multivalent cations and charge shielding by proteins seem to play dominant roles under physiological conditions (**Figure 2**).

PI(4,5)P₂ binding to multivalent cations has been reported in both monolayers and bilayers. Early investigations found cation binding to PI(4,5)P₂ outside the context of membrane bilayers. Ca²⁺ or Mg²⁺ dramatically affected the elution profile of PI(4,5)P₂ through an ion exchange column, probably by decreasing the effective charge of PI(4,5)P₂ when a stable chelate formed (136). Ca²⁺ or Mg²⁺ binding to PI(4,5)P₂ induced large multilamellar PI(4,5)P₂ aggregates (137). Later, in both model membranes and red blood cell membranes, cation binding to PI(4,5)P₂ revealed that PI(4,5)P₂ has higher affinity for Ca²⁺ than for Mg²⁺ (138, 139). Studies of Langmuir monolayers showed that divalent cations bind to PI(4,5)P₂ (140). In contrast to chaotropic agents such as monovalent salts, which expanded the area per PI(4,5)P₂ molecule in monolayers, Ca²⁺ significantly condensed the area per PI(4,5)P₂ molecule. Surface pressure measurements showed that this condensing effect is reversible. The addition of Ca²⁺ decreased the surface pressure in a constant monolayer area. The surface pressure recovered to the normal level upon adding an excess of EDTA to chelate all of the Ca²⁺ (141). Additional studies on membrane bilayers provided more details of PI(4,5)P₂ and multivalent cation association. Based on electrophoretic mobility measurements (21), the intrinsic association constants of Ca²⁺ and Mg²⁺ for PI(4,5)P₂ were 500 M⁻¹ and 100 M⁻¹, respectively. Based on surface potential measurements, another study reported that the calculated intrinsic association constants for PI(4,5)P₂ in bilayers were 360 M⁻¹ for Ca²⁺ and 220 M⁻¹ for Mg²⁺ (141). Attenuated total reflection Fourier transform infrared measurements suggest that upon binding to PI(4,5)P₂ headgroups, partial water loss occurs from the hydration shell of Ca²⁺ but not Mg²⁺, presumably due to the tight binding of Mg²⁺ to water molecules.

Collectively, these studies provide evidence of Ca^{2+} and Mg^{2+} binding tightly to $\text{PI}(4,5)\text{P}_2$ head-groups and indicate cluster promotion occurs upon binding under physiological ionic conditions.

Several groups have visualized divalent cation-induced $\text{PI}(4,5)\text{P}_2$ cluster formation on monolayers, bilayers, or cell membranes. When including fluorescently labeled $\text{PI}(4,5)\text{P}_2$, Ca^{2+} -dependent $\text{PI}(4,5)\text{P}_2$ clusters were seen in monolayers at $\text{PI}(4,5)\text{P}_2$ concentrations between 8 and 50 mol% by using fluorescence microscopy (140). Ellenbroek et al. (142) published $\text{PI}(4,5)\text{P}_2$ phase diagrams in monolayers with 1 mM Ca^{2+} based on atomic force microscopy (AFM) and fluorescence microscopy. They showed that divalent cation-induced $\text{PI}(4,5)\text{P}_2$ clusters were visible at as low as 2 mol% $\text{PI}(4,5)\text{P}_2$ at pH 7.4. Using supported lipid monolayers, the same group later used tapping mode AFM imaging to visualize submicrometer-sized $\text{PI}(4,5)\text{P}_2$ clusters with a radius of approximately 40 nm induced by micromolar Ca^{2+} and smaller clusters induced by millimolar Mg^{2+} (141). A different group found that in the absence of divalent cations, GUVs that contain 5 mol% total $\text{PI}(4,5)\text{P}_2$ and 0.1% fluorescent $\text{PI}(4,5)\text{P}_2$ looked uniform. Divalent cations induced $\text{PI}(4,5)\text{P}_2$ cluster formation in a concentration-dependent manner above 25 μM for Ca^{2+} and above 300 μM for Mg^{2+} . Furthermore, the presence of high concentrations of divalent cations (>300 μM for Ca^{2+} and >1 mM for Mg^{2+}) rendered the GUVs more fragile and caused vesicle rupture (122). However, a different group (143) did not observe $\text{PI}(4,5)\text{P}_2$ clusters in GUVs containing 1 mol% $\text{PI}(4,5)\text{P}_2$ with 100 μM Ca^{2+} . Their fluorescence depolarization data implied that an average cluster has approximately 15 $\text{PI}(4,5)\text{P}_2$ molecules. $\text{PI}(4,5)\text{P}_2$ clusters on cell membrane sheets have been visualized primarily by $\text{PI}(4,5)\text{P}_2$ -specific binding proteins with fluorescent tags, such as PH-PLC δ , or by $\text{PI}(4,5)\text{P}_2$ antibodies, via use of fluorescence microscopic methods. For example, one group (119) used dSTORM to image PC12 cell membrane sheets with anti- $\text{PI}(4,5)\text{P}_2$ antibodies directly conjugated with Alexa Fluor 647. They estimated the $\text{PI}(4,5)\text{P}_2$ cluster size to be ~ 65 nm. As mentioned in Section 6, Honigmann et al. (123) used STED microscopy to find that $\text{PI}(4,5)\text{P}_2$ clusters in PC12 cell membranes had an average diameter of ~ 70 nm when detected with fluorescent PH-PLC δ and an average diameter of ~ 90 nm when detected by a monoclonal $\text{PI}(4,5)\text{P}_2$ antibody plus a secondary antibody labeled with Alexa Fluor 488. In summary, these imaging studies on monolayers, GUVs, and cell membrane sheets indicate that $\text{PI}(4,5)\text{P}_2$ -cation clusters are most likely to be submicroscopic by conventional confocal microscopy. The size of these nanoscopic $\text{PI}(4,5)\text{P}_2$ clusters could be determined only by using superresolution microscopy with optimized labeling methods. It has been possible to create spatial patterns of $\text{PI}(4)\text{P}$ and $\text{PI}(4,5)\text{P}_2$ in confined regions of supported lipid bilayers, in the presence of both PIP5K and 5'-phosphatases, when the reaction was geometrically confined (144). Stochastic geometry sensing might provide a mechanism for creating lipid gradients in confined structures, such as membrane sheets, blebs, vesicles, and tubules. These spatial patterns of $\text{PI}(4)\text{P}$ and $\text{PI}(4,5)\text{P}_2$ were visualized with fluorescent DrrA (a Rab recruitment protein) and PH-PLC δ , respectively. However, the reaction buffer used in this study contained 5 mM Mg^{2+} , which can induce clustering of both $\text{PI}(4)\text{P}$ and $\text{PI}(4,5)\text{P}_2$, even in the absence of enzymes. The influence of fluorescent protein probes like PH-PLC δ or fluorescent lipid antibodies on cluster formation and size should be carefully investigated.

Determining lipid lateral diffusion is crucial for understanding membrane spatial heterogeneity. Another way to confirm $\text{PI}(4,5)\text{P}_2$ clustering is to determine whether $\text{PI}(4,5)\text{P}_2$ molecular diffusion slows down in the membrane. Lipid diffusion has been studied in model membranes and intact cell membranes by methods such as fluorescence correlation spectroscopy (FCS) and fluorescence recovery after photobleaching (FRAP) (145). The Prieto group (143) detected Ca^{2+} -induced clustering of $\text{PI}(4,5)\text{P}_2$ with FCS. The diffusion coefficient of $\text{PI}(4,5)\text{P}_2$ decreased in the presence of 100 μM Ca^{2+} , while that of the control lipid POPS (palmitoyl

oleoyl phosphatidylserine) remained unchanged. This study concluded that the high sensitivity of PI(4,5)P₂ diffusion to Ca²⁺ indicates that PI(4,5)P₂ cluster size must be larger than dimers. The Janmey group (141) also used FCS to detect cation-induced slowing of PI(4,5)P₂ diffusion in GUVs. The presence of 1 mM Mg²⁺ slowed the diffusion of fluorescently labeled PI(4,5)P₂ by fourfold, consistent with nanoscopic PI(4,5)P₂ clusters being induced by Mg²⁺. Compared with Mg²⁺ and Zn²⁺, Ca²⁺ slowed PI(4,5)P₂ diffusion more. The McLaughlin group (73) reported that the diffusion of PI(4,5)P₂ is 10 times slower in the presence of micromolar Ca²⁺. This study also showed that in the presence of 1 mol% PI(4,5)P₂ in GUVs, the diffusion rate of a membrane-associated fluorescently labeled polybasic peptide, Lys13, is only approximately half of that in a PI(4,5)P₂-free membrane, indicating that the polybasic peptide diffuses together with the laterally sequestered PI(4,5)P₂. This result suggests that there can be two different PI(4,5)P₂ populations present in the PM. Another paper from this group (146) further explored PI(4,5)P₂ diffusion in the PM of fibroblasts and epithelial cells, as well as in model membranes like GUVs. They used fluorescent PI(4,5)P₂ to label either the outer leaflet by incubating GUVs with PI(4,5)P₂-containing micelles or the inner leaflet by microinjecting PI(4,5)P₂-containing micelles into GUVs. They reported that PI(4,5)P₂ diffusion at the cellular inner leaflet of the PM was two- to fourfold slower than the PI(4,5)P₂ diffusion at the cellular outer leaflet of the PM, in PM-derived blebs, and on GUVs. The lower diffusion coefficient of PI(4,5)P₂ in the inner leaflet is consistent with the hypothesis that approximately two thirds of the PI(4,5)P₂ is reversibly and electrostatically associated with multivalent cations and basic membrane proteins or the cytoskeleton, while the rest of the PI(4,5)P₂ is free. Together, these studies provide strong evidence that the PI(4,5)P₂ diffusion rate slows down significantly when PI(4,5)P₂ cluster formation occurs as a result of multivalent cation bridging and sequestration by cellular proteins.

Since PI(4,5)P₂-cation clusters are most likely to be nanoscopic, biophysical methods such as FRET, fluorescence self-quenching, and fluorescence anisotropy are required to provide molecular details about these clusters. The Janmey group (141) used FRET to study divalent cation-induced PI(4,5)P₂ clustering in membrane bilayers and found that the trend in inducing PI(4,5)P₂ cluster formation follows the order of Ca²⁺ ≫ Mg²⁺ > Zn²⁺ at a fixed 5 mol% PI(4,5)P₂. Using fluorescence anisotropy, the Prieto group (143) reported that Ca²⁺ at concentrations up to 100 μM induces PI(4,5)P₂ clusters. At identical concentrations, Ca²⁺ is significantly more efficient than Mg²⁺ at promoting PI(4,5)P₂ clustering (147), but 5 mM Mg²⁺ was able to drive PI(4,5)P₂ clustering more efficiently than 100 μM Ca²⁺. The Prieto group (143) also used fluorescence self-quenching to investigate the PI(4,5)P₂ lateral distribution in PC bilayers. The steady-state fluorescence intensity of TopFluor-PI(4,5)P₂ was measured at several concentrations. Self-quenching was observed as Ca²⁺ concentration increased, with no self-quenching seen in the presence of 5 mM EDTA for up to 1 mol% total PI(4,5)P₂ (143). Most previous studies focused only on the effect of physiological cations such as Ca²⁺ with the Ca²⁺-free condition as the negative control. However, without adding the metal chelators EDTA or EGTA to buffers, or without using CHELEX-treated buffers, other multivalent cations are likely to be present (148). Discrepancies in the PI(4,5)P₂ literature come from multivalent cation contamination in buffers, as we showed recently (131). Thus, careful sample and buffer preparation is necessary to study PI(4,5)P₂ behavior.

Although PI(4,5)P₂ clustering is supported by abundant evidence, most previous *in vitro* studies used nonphysiological PI(4,5)P₂ concentrations greater than 5 mol% and buffers containing uncharacterized multivalent cation contamination. Our recent study (131) showed that buffers prepared with American Chemical Society-grade chemicals with 99% purity and stored in borosilicate glass bottles contained Ca²⁺ (60 μM), Al³⁺ (10 μM), Zn²⁺ (6 μM), and Fe³⁺ (0.1 μM) by measurement with inductively coupled plasma optical emission spectroscopy.

Characterizing PI(4,5)P₂ behavior at high PI(4,5)P₂ concentrations misses the origin of PI(4,5)P₂–PI(4,5)P₂ association. Recently, we found a maximal solubility of free PI(4,5)P₂ in model membranes, above which PI(4,5)P₂ clustering occurs in the presence of multivalent cations. Both self-quenching and FRET were used to examine PI(4,5)P₂ self-aggregation behavior over a 200-fold concentration range from 0.01 to 2 mol% in model bilayers that mimic the inner leaflet lipid composition, including PE, PS, and cholesterol, in buffers with known cation types and concentrations. PI(4,5)P₂ starts to self-associate at extremely low concentrations of 0.02–0.05 mol% of total lipids, with multivalent cations absolutely required for this type of PI(4,5)P₂ clustering. The maximal PI(4,5)P₂ solubility prior to clustering was termed the critical PI(4,5)P₂ concentration (CPC) by analogy with the CMC. This type of PI(4,5)P₂ association was described to be PI(4,5)P₂–cation clustering because EDTA eliminates the PI(4,5)P₂ clusters. For both FRET and self-quenching, the linear increase with increasing PI(4,5)P₂ above the CPC implies that all additional PI(4,5)P₂ forms more clusters that possess the same properties. This abrupt onset of PI(4,5)P₂ self-association is a characteristic of high-order aggregation, as seen in micelle formation or phase separation. PI(4,5)P₂–cation cluster formation is not dependent on a particular multivalent cation. Al³⁺ had the strongest effect on cluster formation and Mg²⁺ the weakest, but a variety of cations promoted clustering, including Ca²⁺, Zn²⁺, and Fe³⁺. As low as 1 μM of Al³⁺, Ca²⁺, Fe³⁺, or Zn²⁺ or 50 μM Mg²⁺ was enough to cause PI(4,5)P₂ clustering above the CPC in a buffer containing 100 mM KCl and 20 mM HEPES. Many previous studies did not consider the possible multivalent cation contaminants present in the micromolar range in salts and buffers. Moreover, metal ions are leached from glass and plastic over time, complicating the interpretation of some previous reports on PI(4,5)P₂ behavior.

We pointed out that PI(4,5)P₂ clustering is a headgroup-specific behavior that is not influenced by the PI(4,5)P₂ hydrophobic acyl chain types (131). Another important finding is that other PIP species can cocluster with PI(4,5)P₂, but PI cannot. This result indicates that the phosphate groups at positions 3, 4, or 5 of the inositol ring are required for the coclustering, with positively charged multivalent cations bridging the negatively charged phosphomonoesters. Thus, pools of different PIPs could exist; for example, PI(4)P and PI(4,5)P₂ could coexist within the same cluster at the inner leaflet of the PM. While multivalent cations are absolutely required for PI(4,5)P₂–cation cluster formation, the bulk membrane composition surrounding the PI(4,5)P₂ also influences PI(4,5)P₂ clustering. For example, self-quenching of fluorescent TopFluor–PI(4,5)P₂ in PE/PI/cholesterol bilayers is sixfold higher than in PC bilayers. Cholesterol or PI promotes even more clustering, though the mechanism is unclear. Physiological levels of 0.5 mM Mg²⁺ that mimic the resting state in cells are sufficient to induce physiological levels of 2 mol% PI(4,5)P₂ to cluster; an additional 100 μM Ca²⁺, mimicking local transient cellular calcium influx, causes even stronger PI(4,5)P₂ clustering. In summary, this study sheds light on how PI(4,5)P₂ exists as spatially separated pools in cells, because it quantitatively pinpoints several experimental factors that affect clustering *in vitro*.

8. HOW PROTEINS RECOGNIZE AND RESPOND TO PI(4,5)P₂ CLUSTERS

In the cytosol, the total concentration of PI(4,5)P₂-binding proteins far exceeds that of PI(4,5)P₂ at the inner leaflet, implying that PI(4,5)P₂ clustering could be highly regulated. In this review, we show that four properties of this unique lipid are key to its behavior (**Figure 3**): (a) Both free and clustered forms of PI(4,5)P₂ can coexist in the same membrane, (b) the lateral distribution of PI(4,5)P₂ is regulated by a variety of bound molecules, (c) different proteins preferentially bind to

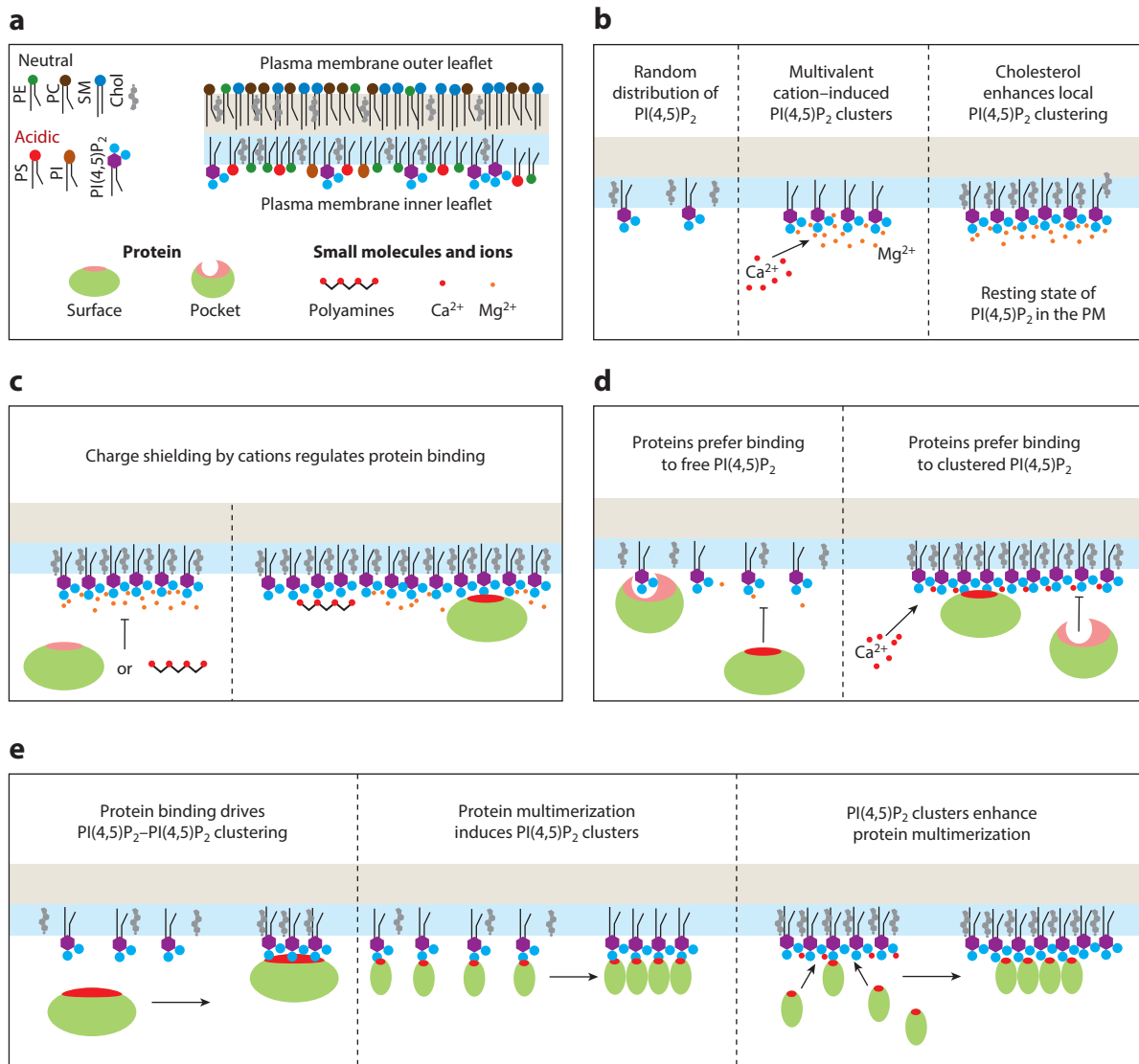


Figure 3

Different modes of protein response to PI(4,5)P₂ clusters. (a) The asymmetric lipid distribution across the PM. (b) PI(4,5)P₂ clusters can form in the presence of physiological multivalent cations. Spatially separated pools of free and clustered PI(4,5)P₂ coexist at the inner leaflet of the PM. (c) Charge shielding and sequestration of negatively charged PI(4,5)P₂ headgroups by cations and polybasic small molecules regulate protein binding to PI(4,5)P₂. (d) Some proteins prefer to bind free PI(4,5)P₂, while others bind PI(4,5)P₂ in clusters. During transient calcium influx or any local cation changes, proteins targeting to free PI(4,5)P₂ could have reduced binding, while proteins targeting to clustered PI(4,5)P₂ could have enhanced binding. (e) Upon membrane binding, some proteins can sequester local PI(4,5)P₂. Some proteins induce PI(4,5)P₂ cluster formation as they multimerize on membranes; in turn, multimerized proteins have enhanced membrane binding. Abbreviations: Chol, cholesterol; PC, phosphatidylcholine; PE, phosphatidylethanolamine; PI, phosphatidylinositol; PI(4,5)P₂, phosphatidylinositol 4,5-bisphosphate; PM, plasma membrane; PS, phosphatidylserine; SM, sphingomyelin.

free and/or clustered PI(4,5)P₂, and (d) multimerization of PI(4,5)P₂-bound proteins can further modulate PI(4,5)P₂ clustering properties.

8.1. Both Free and Clustered Forms of PI(4,5)P₂ Can Coexist in the Same Membrane

Separated pools of free and clustered PI(4,5)P₂ can coexist in the same membrane (**Figure 3a,b**). PI(4,5)P₂-cation clusters form at extremely low concentrations within model membranes, a new behavior of PI(4,5)P₂ that we propose also occurs within cell membranes (131). PI(4,5)P₂ clusters can form by local multivalent cations bridging PI(4,5)P₂ headgroups. In a cellular resting state, there is some free PI(4,5)P₂, but most is clustered PI(4,5)P₂ due to the presence of Mg²⁺ and binding proteins. During transient Ca²⁺ influx, local PI(4,5)P₂ clustering is enhanced. Moreover, PI(4,5)P₂ clustering can be influenced by the local lipid environment; for example, cholesterol seems to enhance local PI(4,5)P₂ clustering. In the presence of PI(4,5)P₂-interacting proteins, protein-induced PI(4,5)P₂ clusters could also form (149). All of these local factors collectively contribute to these spatially separated pools of free, mildly clustered, or tightly clustered PI(4,5)P₂, which could further regulate protein-binding events.

8.2. The Lateral Distribution of PI(4,5)P₂ Is Regulated by a Variety of Bound Molecules

PI(4,5)P₂ cluster-inducing binders such as multivalent cations, polyamines, and proteins with basic patches regulate PI(4,5)P₂ properties and further regulate other cellular factors having access to PI(4,5)P₂ (**Figure 3c**). Although the PI(4,5)P₂ concentration in the PM cytosolic leaflet is 1–2 mol%, a majority of this PI(4,5)P₂ must be sequestered by proteins or by multivalent cations. The lateral organization and function of PI(4,5)P₂ could undergo significant changes in the presence of physiologically relevant concentrations of multivalent cations, which influence PI(4,5)P₂-dependent cellular processes. One study (150) found that intracellular Mg²⁺ inhibits a PI(4,5)P₂-requiring ion channel called KCNQ2/3. Mg²⁺ reduced the current by electrostatically binding to PI(4,5)P₂, significantly reducing the free PI(4,5)P₂ available for interaction with these channels. This mechanism is likely to modulate many other PI(4,5)P₂-dependent ion channels and cellular processes. In another example, Mg²⁺ and polyamines electrostatically bind to PI(4,5)P₂ and could inhibit the activity of PLC by decreasing the amount of free PI(4,5)P₂ available for hydrolysis (151). The homeostasis of metal cations is modulated by different transporters (152). The transient influx upon signaling activation by metal ions such as Ca²⁺ or Zn²⁺ could potentially regulate PI(4,5)P₂ lateral distribution in the membrane and accessibility to other PI(4,5)P₂-binding proteins. The details of electrostatic charge shielding of PI(4,5)P₂ by multivalent cations, polyamines, and polybasic proteins would differ based on the membrane charge density, ionic radius, dehydration energy of cations, and manner of bringing together multiple charges. However, one final result could be similar: Shielding and sequestering negatively charged PI(4,5)P₂ headgroups prevent PI(4,5)P₂ from interacting with certain binding molecules (**Figure 3c**).

8.3. Different Proteins Preferentially Bind to Free and/or Clustered PI(4,5)P₂

Some proteins prefer to bind free PI(4,5)P₂ while others prefer to bind PI(4,5)P₂ in clusters (**Figure 3d**). Several, or even many, populations of PI(4,5)P₂ could coexist depending on local multivalent cation concentration and local lipid concentration, and free PI(4,5)P₂ could be in equilibrium with clustered PI(4,5)P₂. The headgroup of free PI(4,5)P₂ is exposed, and thus

proteins that preferentially bind to sparsely distributed free PI(4,5)P₂ would have immediate access. The headgroup of clustered PI(4,5)P₂ must be bound by cations or proteins; those proteins that preferentially bind to clustered PI(4,5)P₂ would be competing against bound cations for access to PI(4,5)P₂ molecules. PI(4,5)P₂-cation clusters could form a two-dimensional array of the many PI(4,5)P₂ phosphate groups together with multivalent cations. The degree of PI(4,5)P₂-cation clustering would depend on the type and concentration of multivalent cations and on the surrounding lipid compositions. Local changes to multivalent cations and surrounding lipids affect the free and clustered PI(4,5)P₂ concentrations, thus further influencing PI(4,5)P₂-binding protein types, concentrations, and functions. For example, during a transient calcium influx, more cation-bridged PI(4,5)P₂ clusters would form, and the proteins that target clustered PI(4,5)P₂ could experience enhanced function, while proteins that target free PI(4,5)P₂ could experience decreased function (**Figure 3d**). Our recent study shows that a fluorescent version of PH-PLC δ 1 exhibits robust binding to free PI(4,5)P₂ on GUVs but reduced binding to clustered PI(4,5)P₂ bridged by multivalent cations (149). The stronger the degree of PI(4,5)P₂ clustering on GUVs, the less overall membrane binding of PH-PLC δ 1.

Several reasons might explain why proteins bind preferentially to free PI(4,5)P₂ on the one hand or to clustered PI(4,5)P₂ on the other. PH-PLC δ 1 and similar proteins contain a canonical PI(4,5)P₂ binding pocket that is proposed to interact with the PI(4,5)P₂ headgroup with 1:1 stoichiometry in a noncooperative manner (57). Multivalent cations binding to PI(4,5)P₂ headgroups are reported to confine the lipid headgroup tilt angle, which based on molecular dynamics simulations would inhibit PH-PLC δ 1-domain recognition (153). Unlike PH-PLC δ 1 with its canonical PI(4,5)P₂-binding site, proteins with a polybasic region or patch, such as MARCKS and N-WASP, bind to PI(4,5)P₂ in a multivalent manner. The activation of N-WASP has been shown to depend on PI(4,5)P₂ concentration; once the PI(4,5)P₂ level is above a threshold, N-WASP binds to PI(4,5)P₂ cooperatively with an apparent Hill coefficient of approximately 20. This sharp PI(4,5)P₂ sensitivity can be tuned by modulating the polybasic region (154). We recently reported that naturally myristoylated HIV-1 MA preferentially binds to clustered PI(4,5)P₂ over free PI(4,5)P₂, while the nonmyristoylated mutant strongly binds to free PI(4,5)P₂ (149). The naturally nonmyristoylated RSV MA also preferentially binds to free PI(4,5)P₂. However, the forced hexameric version of RSV MA exhibits the opposite binding profile, preferring binding to clustered PI(4,5)P₂ over free PI(4,5)P₂. From these results we propose that only proteins with strong membrane affinity can compete with PI(4,5)P₂-shielding multivalent cations to gain access to PI(4,5)P₂ headgroups, while proteins with weak membrane affinity have access to free PI(4,5)P₂ headgroups but not to headgroup-shielded PI(4,5)P₂.

8.4. Multimerization of PI(4,5)P₂-Bound Proteins Can Further Modulate PI(4,5)P₂ Clustering Properties

Clustering of PI(4,5)P₂-bound proteins could sequester PI(4,5)P₂ or even induce PI(4,5)P₂ clustering (**Figure 3e**), especially if those proteins can multimerize on membranes. Clustered PI(4,5)P₂ could then recruit more proteins to achieve downstream functions. The behavior of syntaxin-1 in PC12 cells (120) shows that clustering of membrane-bound proteins can lead to formation of PI(4,5)P₂ clusters by electrostatic interactions with juxtamembrane basic residues. Another example of protein multimerization leading to PI(4,5)P₂ clusters involves the BAR superfamily of proteins. The I-BAR, F-BAR, and N-BAR domains from many mammalian proteins have been shown to induce PI(4,5)P₂ clustering (155, 156). These BAR domains assemble into stable scaffolds to bend membranes and restrict PI(4,5)P₂ lateral diffusion by generating stable PI(4,5)P₂-BAR domain clusters. Besides PI(4,5)P₂, BAR domains could also cluster PI(3,4,5)P₃

and phosphatidylinositol 3-phosphate (PI3P), indicating that this clustering occurs through electrostatic sequestration. Several studies show that the PH domain of dynamin-1 and -2 binds to PI(4,5)P₂ with low affinity and a K_d value above 1 mM (157). However, oligomerization of dynamin can locally cluster PI(4,5)P₂ at coated pits. Concentrated PI(4,5)P₂ allows multiple PH domains to interact with membranes with higher affinity, promoting vesicle scission during CME (158, 159).

Similar to clustering by cellular proteins, PI(4,5)P₂ sequestration can be induced by viral proteins. PI(4,5)P₂ enrichment in HIV-1 viral membranes and the efficient self-assembly of the Gag protein have led to competing, but not mutually exclusive, hypotheses: First, PI(4,5)P₂-rich domains preexist and are targeted by Gag; second, PI(4,5)P₂ clustering is induced by Gag multimerization during assembly (160, 161). Our recent study provides comprehensive evidence that Gag assembly on membranes induces PI(4,5)P₂ clusters (149). Both specific PI(4,5)P₂ interactions and Gag–Gag interactions are required for this protein-induced PI(4,5)P₂ clustering effect, as the relevant mutants fail to cluster PI(4,5)P₂. Protein-induced PI(4,5)P₂ clustering and multivalent cation-induced PI(4,5)P₂ clustering are independent and additive. Collectively, we show that HIV-1 Gag not only targets preexisting PI(4,5)P₂ clusters but also further enriches PI(4,5)P₂ at assembly sites while Gag multimerizes. PI(4,5)P₂ clustering and protein multimerization are likely to facilitate each other, creating membrane platforms for optimal protein function.

9. CONCLUSIONS AND OUTSTANDING QUESTIONS

The rich physical chemistry of PI(4,5)P₂ contributes to its being an interesting and important phospholipid in cells. Despite electrostatic repulsion, PI(4,5)P₂ can form clusters via two major attractive interactions—hydrogen bonding and, especially, multivalent cation bridging, with the latter likely having the predominant role in cells. Spatially separated pools of PI(4,5)P₂ could arise from localized PI(4,5)P₂ aggregation due to local differences in multivalent cations and lipid compositions as well as to localized PI(4,5)P₂ synthesis and breakdown. PI(4,5)P₂ clustering changes the orientation and accessibility of the headgroups, and this has many implications for the biological function of this phospholipid. A given protein is likely to preferentially bind to either free PI(4,5)P₂ or clustered PI(4,5)P₂; PI(4,5)P₂-bound proteins are also likely to either promote or disrupt preexisting PI(4,5)P₂ clusters. Tight control of local synthesis and turnover of PI(4,5)P₂, together with well-regulated cluster formation and recruitment of selected proteins to specific sites at the inner leaflet of the PM, create a broad spectrum of PI(4,5)P₂-dependent cellular functions.

PI(4,5)P₂ clusters likely exist in all cells and are recognized and utilized by hundreds of cellular proteins, so understanding the nature of PI(4,5)P₂ behavior is fundamental to cell biology. It is of great importance to further investigate the biophysical properties of PI(4,5)P₂-cation and PI(4,5)P₂-protein clusters, especially upon cellular protein binding, transient calcium influx, or zinc sparks. The last is a fertilization event that induces the exocytotic release of billions of zinc ions (162). Only when we have a better understanding of this unique lipid will we be able to seek answers for more complicated questions. In the future, it is critical to elucidate the mechanisms by which PI(4,5)P₂, an inner leaflet lipid, is mysteriously found associated with rafts in PM outer leaflet microdomains. Many enveloped viruses other than HIV-1 likely also rely on PI(4,5)P₂ clusters to assemble progeny viral particles and then bud from infected cells; perhaps this specific PI(4,5)P₂ dependence could be these viruses' Achilles' heel.

SUMMARY POINTS

1. Diverse cellular and viral proteins interact with PI(4,5)P₂ using different binding strategies, such as surface electrostatic interactions or a specific binding pocket.
2. PI(4,5)P₂ starts to self-associate at extremely low concentrations in membrane bilayers, forming nanoscopic PI(4,5)P₂ clusters with headgroups bridged by multivalent cations.
3. Other PIP species are able to cocluster with PI(4,5)P₂ due to positively charged multivalent cations bridging the negatively charged phosphomonoesters.
4. Differences in local multivalent cation types and concentrations, as well as the surrounding lipid environments, could significantly influence local PI(4,5)P₂ clustering, resulting in spatially separated pools of free and clustered PI(4,5)P₂.
5. Electrostatic shielding and sequestration of PI(4,5)P₂ headgroups by multivalent cations and proteins with basic patches regulate PI(4,5)P₂ accessibility to other PI(4,5)P₂ binders.
6. Some cellular and viral proteins preferentially bind to free PI(4,5)P₂, some preferentially bind to clustered PI(4,5)P₂, and others do not have a preference.
7. Proteins binding to PI(4,5)P₂, especially those that can multimerize, can laterally sequester PI(4,5)P₂ or induce PI(4,5)P₂ clustering electrostatically.

DISCLOSURE STATEMENT

The authors are not aware of any affiliations, memberships, funding, or financial holdings that might be perceived as affecting the objectivity of this review.

ACKNOWLEDGMENTS

We thank Robert A. Dick for figure preparation, discussions, and editing. This work was supported by the National Institutes of Health (awards R01-AI150454 to V.M.V., R01-GM1045684 to G.W.F., and R01-AI147890 to Robert A. Dick).

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